

# Separation and Purification of Four Stilbenes from *Vitis vinifera* L. cv. Cabernet Sauvignon Roots Through High-speed Counter-current Chromatography

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Abbreviations: HSCCC, high-speed counter-current chromatography

HPLC, high performance liquid chromatography

NMR, nuclear magnetic resonance

**A method for preparative separation and purification of *trans*-resveratrol,  $\delta$ -viniferin,  $\epsilon$ -viniferin and *trans*-vitisin B from the roots of *Vitis vinifera* L. cv. Cabernet Sauvignon was successfully established and is reported on in this paper. The four important stilbenes were purified by high-speed counter-current chromatography (HSCCC) with a suitable quaternary solvent system composed of chloroform–methanol–*n*-butanol–water (4:3:0.05:2, v/v). A total of 7.1 mg  $\pm$  0.2 mg of *trans*-resveratrol, 1.1 mg  $\pm$  0.1 mg of  $\delta$ -viniferin, 18.7 mg  $\pm$  0.5 mg of  $\epsilon$ -viniferin, and 12.2 mg  $\pm$  0.2 mg of *trans*-vitisin B, with purities of 97.89%, 90.61%, 94.37% and 78.38% respectively, were obtained from 241 mg of crude sample in a one-step HSCCC separation. The chemical structures of *trans*-resveratrol and  $\delta$ -viniferin were further confirmed with the retention time using the method of standard addition, while the structural identification of  $\epsilon$ -viniferin and *trans*-vitisin B was performed with LC-ESI/MS, <sup>1</sup>H-NMR, and <sup>13</sup>C-NMR.**

## INTRODUCTION

Stilbenes, an important subclass of polyphenolic compounds (Guebailia *et al.*, 2006) that include resveratrol, resveratrol glycoside, methylated resveratrol and viniferins, are phytoalexins found in grapes and other food products (Jang *et al.*, 1997). All resveratrol oxidation products, which consist of dimers, trimers and the more highly polymerised oligomers of resveratrol, are known as viniferins. The bioactivities of resveratrol glycoside and viniferins have been studied less than resveratrol because their concentrations in grapes and wine are lower than those of resveratrol; however, some of their health-promoting properties have been found (Murias *et al.*, 2005; Kuo & Hsu, 2008; Guerrero *et al.*, 2009). Extensive studies have shown that stilbenes have several useful biological properties, such as anti-inflammatory, anti-asthmatic (Lee *et al.*, 2009), anti-cancer (Jang *et al.*, 1997), anti-HIV (Bedoya *et al.*, 2006), anti-cardiovascular diseases (Ong *et al.*, 2011), hepatoprotective (Lee *et al.*, 2012) and anti-depressant activities (Yáñez *et al.*, 2006), and that they protect against ethanol-induced oxidative DNA damage (Yan *et al.*, 2012) and amyloid

fibril formation (Rivière *et al.*, 2010). Grapes are one of the most important potential sources of resveratrol for humans. Grapes (genus *Vitis* of family Vitaceae) are rich sources of stilbenes, especially resveratrol oligomers (He *et al.*, 2009). Resveratrol and resveratrol glycoside compounds are often reported in wine (Bavaresco *et al.*, 2003; Paulo *et al.*, 2011). Paulo *et al.* (2011) analysed 186 Portuguese red wines and found that their *trans*-resveratrol content ranged from 0.05 mg L<sup>-1</sup> to 10.9 mg L<sup>-1</sup> (Paulo *et al.*, 2011). However, some studies have shown that the types of stilbenes in wine differ and that their content is less than in grape stems and roots (Pezet *et al.*, 1994; Wang *et al.*, 2010; Vergara *et al.*, 2012). Thus, grape stems and roots are potential sources of stilbenes for separation and purification.

HSCCC is a support-free liquid–liquid partition chromatography, in which both the mobile and the stationary phase are liquids. HSCCC has significant advantages over other special separation techniques, which eliminate the irreversible adsorptive sample loss, stationary phase deactivation and tailing of solute peaks because of the surface overloading of silanol (Yao *et al.*, 2012). In addition,

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HSCCC can handle large sample loads, has high recovery, and is scalable (Fang *et al.*, 2011). The closed separation system protects the thermosensitive and photosensitive bioactive samples; hence, it is well suited for studying bioactive natural products. In recent years, HSCCC has been widely used in research on and the development of natural medicines and functional foods, including phytosterols (Schröder & Vetter, 2011), minor saponins (Ha *et al.*, 2011), essential oils (Chen *et al.*, 2011), epigallocatechin gallate (Du *et al.*, 2000), gingerols and 6-shogaol (Qiao & Du, 2011), neohesperidin (Zhang *et al.*, 2012), *Alpinia katsumadai hayata* flavonoids (Xiao *et al.*, 2011), tanshinone (Tian *et al.*, 2000; Sun *et al.*, 2011), ginsenosides (Shehzad *et al.*, 2012), and 6,7-dimethoxycoumarin (He *et al.*, 2012). The chloroform–methanol–water solvent system was used to isolate resveratrol (Chen *et al.*, 2001), the cyclohexane–ethyl acetate–methanol–water solvent system was used to isolate stilbene glucosides (Fan *et al.*, 2009), *n*-hexane–ethyl acetate–methanol–water solvent system was used to isolate hopeaphenol, amurensin G and vitisin A (He *et al.*, 2009); and the ethyl acetate–ethanol–water solvent system was used to isolate stilbene glucoside and catechin (Jin & Tu, 2005).

The purpose of this work was to establish a very convenient and efficient method for the rapid separation of stilbenes from the roots of *Vitis vinifera* L. cv. Cabernet Sauvignon. Firstly,

four stilbenes from the roots of grapevines were isolated by HSCCC and semi-preparative high-performance liquid chromatography. Secondly, the *trans*-resveratrol and the  $\delta$ -viniferin were identified via UV detection and compared with standards. The chemical structures of  $\epsilon$ -viniferin and *trans*-vitisin B (Fig. 1) were positively identified via UV detection, LC-ESI-MS/MS,  $^1\text{H}$  NMR and  $^{13}\text{C}$  NMR. This is the first report on the separation of four stilbenes from the roots of *Vitis vinifera* L. cv. Cabernet Sauvignon using this combination of techniques.

## MATERIALS AND METHODS

### Reagents and materials

All solvents used to prepare the crude samples and separate the HSCCC were of analytical grade and obtained from Beijing Beihua Fine Chemicals Company (Beijing, China). Chromatography grade solvents were purchased from J & K Technology Co., Ltd. (Beijing, China). The water was prepared using a Milli-Q system (Molsheim, France). The *trans*-resveratrol standard was purchased from Sigma-Aldrich (Sigma-Aldrich, St. Louis, MO). The  $\delta$ -viniferin standard was prepared in our laboratory and its chemical structure was identified via UV, LC-ESI-MS/MS and NMR and compared with a reference (Pezet *et al.*, 2003).

Two-year-old roots of *V. vinifera* L. Cabernet Sauvignon

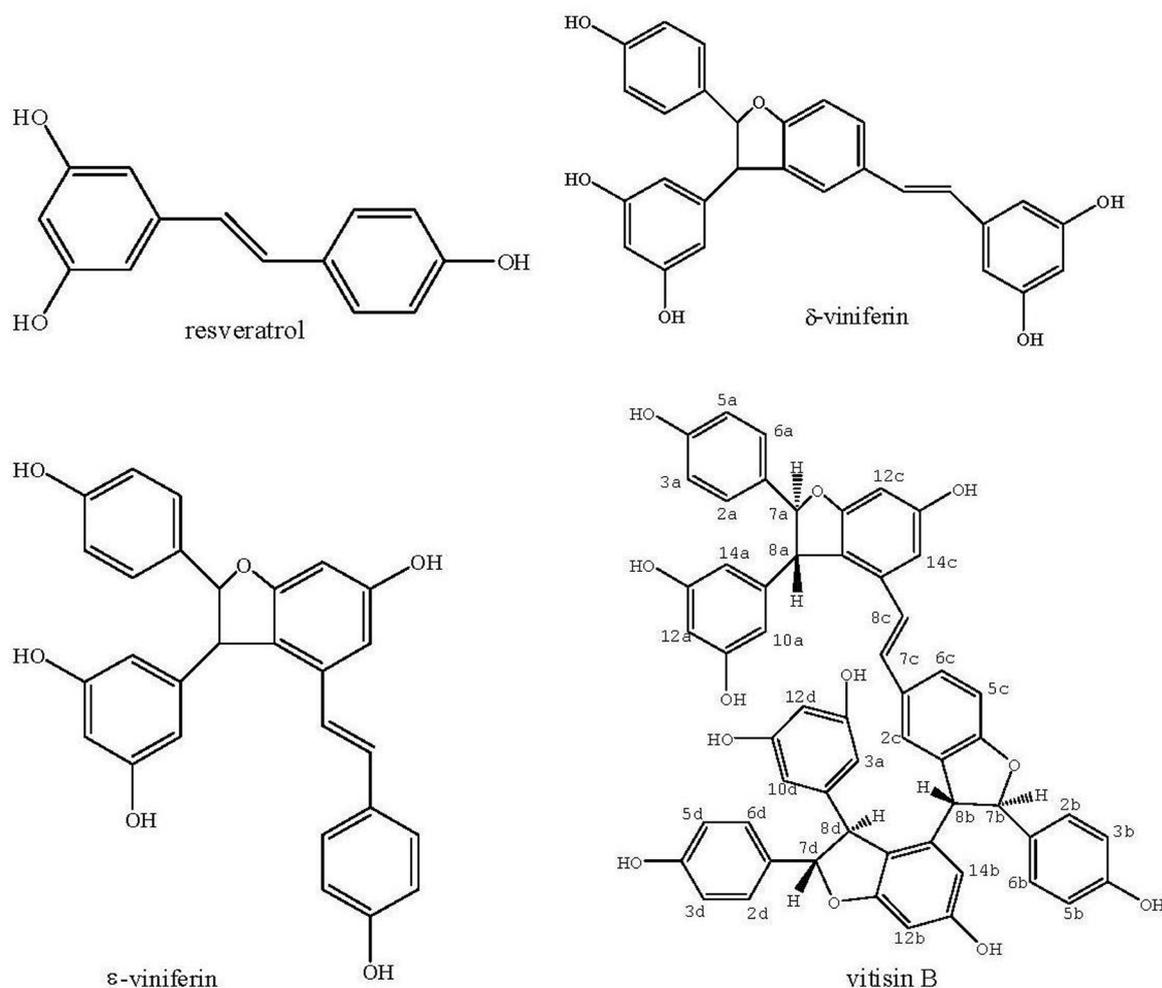


FIGURE 1  
Chemical structures of *trans*-resveratrol,  $\delta$ -viniferin,  $\epsilon$ -viniferin and *trans*-vitisin B.

were collected from the Shangzhuang Experiment Station (China Agricultural University, Beijing, China).

### Apparatus

A TBE-300B HSCCC (Shanghai Tauto Biotech Co., Ltd., Shanghai, China) equipped with three multilayer polytetrafluoroethylene coil separation columns (1.6 mm tube diameter, 280 mL total column volume) and a 20 mL sample injection loop was used in this study. The revolution radius was 5 cm and the resulting  $\beta$ -value of the multilayer coil ranged from 0.5 at the internal terminal to 0.8 at the external terminal ( $\beta = r/R$ , where  $r$  is the distance from the coil to the holder shaft). The HSCCC system was equipped with a TBP5002 constant flow pump (Shanghai Tauto Biotech Co., Ltd., Shanghai, China) and a 313 nm TBD-2000 UV detector (Shanghai Tauto Biotech Co., Ltd., Shanghai, China). The separation temperature was controlled using an HX-105 water-circulating constant temperature implement (Beijing Changliu Scientific Instruments Company, Beijing, China). The chromatogram was recorded using a WH500-USB workstation (Tauto Biotechnology Company, Shanghai, China). The HPLC equipment was a Shimadzu LC-20AT system with an LC-20AT solvent delivery unit, an SPD-20A detector, a Rheodyne 7725i injection valve with a 20  $\mu$ L loop, and LabSolutions/LCSolution software (Shimadzu, Tokyo, Japan). The ESI mass spectra were obtained using an Agilent 6460 Triple Quadrupole HPLC-ESI-MS/MS instrument (Agilent Technologies, USA). The proton and carbon NMR and HSQC spectra were recorded on a Bruker Avance 600 spectrometer (Germany), with  $\text{Me}_4\text{Si}$  as internal standard, in  $\text{CD}_3\text{OD}$ .

### Preparation of the crude sample solution

About 100 g of the freeze-dried root powder (20 mesh to 30 mesh) of *V. vinifera* L. Cabernet Sauvignon was extracted twice for 12 min with 1 L of methanol-water solution under ultrasonication (80 kHz, < 30°C) (70:30, v/v) per extraction, and then placed in a refrigerator protected from light for 1 h at 4°C. Both extracted solutions were combined and evaporated using a rotary evaporator at 35°C under vacuum. The residue was suspended in 300 mL of 3%  $\text{NaHCO}_3$  (w/v) and extracted three times with 200 mL of ethyl acetate. The extracts were then pooled and evaporated using a rotary evaporator at 35°C under vacuum. The 1.8691 g crude extract was stored at -40°C for the subsequent experiments.

### HSCCC

#### Preparation of the two-phase solvent system

The two-phase solvent system was selected according to the partition coefficients ( $K$ ) of each stilbene. The  $K$  values were determined via HPLC as follows: 2 mg of crude extract was dissolved in 2 mL of pre-equilibrated solvent system (upper phase/lower phase, 1:1 v/v). The tube was then shaken vigorously for 3 min. After equilibration, the stilbene content (500  $\mu$ L) of the upper phase and the lower phase was analysed for the target compounds, and the areas of their HPLC peaks were recorded as  $A_{\text{upper}}$  and  $A_{\text{low}}$  respectively. The  $K$  value was calculated as follows:  $K = A_{\text{upper}}/A_{\text{low}}$ .

### HSCCC separation procedure

A two-phase solvent system composed of chloroform, methanol, *n*-butyl alcohol and water (4:3:0.05:2, v/v) was selected to separate the stilbenes from the crude *V. vinifera* extract. The column was first filled entirely with the upper phase, which was used as the stationary phase. After the column was filled with the upper phase, the lower phase was pumped into the column from top to bottom at a flow rate of 2.5 mL/min, while the apparatus was rotated at 900 rpm. After reaching hydrodynamic equilibrium, 10 mL of the sample solution (24 mg/min) was injected into the column using an injection valve and signal acquisition was started. The effluent from the tail end of the column was continuously monitored through UV detection at 313 nm. The separated peak fractions were collected manually according to the chromatographic profiles. All experiments were completed at room temperature.

### Semi-preparative HPLC purification

The liquid chromatography (LC) system consisted of a pump (LC-20 AT) and a diode array detector (SPD-M20A), both purchased from Shimadzu (Shimadzu, Tokyo, Japan). Data acquisition was performed by LabSolutions/LCSolution software (Shimadzu, Tokyo, Japan). The analytical column was a Venusil XBP C18 (250 mm  $\times$  10 mm, 5  $\mu$ m; Agela Technologies, Tianjin, China). The detector wavelength setting was 306 nm. An isocratic elution method with 40% acetonitrile solution (v/v) at a flow rate of 3 mL/min was used. The peak fractions were collected manually according to the chromatographic profiles.

### Identification of HSCCC peak fractions

The extracts and purified compounds were analysed on a Pinnacle II C18 column (250 mm  $\times$  4.6 mm, 5  $\mu$ m; Restek, USA). The mobile phase consisted of ultrapure water (A) and acetonitrile (B). The elution was performed using the following gradient profile: 20% B keep 5 min; 5 min to 21 min, 20% to 38% B; 21 min to 38 min, 38% to 50% B; 38 min to 42 min, 50% to 85% B; 42 min to 45 min, 85% to 20% B; and held until 50 min. The flow rate of the mobile phase was 1 mL  $\text{min}^{-1}$ .

During the ESI-MS analysis, the MS was operated in multiple reaction monitoring mode. Source parameter: gas temperature 325°C; gas flow 6 L  $\text{min}^{-1}$ ; nebuliser 45 psi; sheath gas temperature 350°C; sheath gas flow 11 L/min; capillary positive 3000 V, negative 3000 V; nozzle positive 500 V, negative 1000 V; collision energy, *trans*-resveratrol 25V/15V,  $\delta$ -viniferin 14V/22V,  $\epsilon$ -viniferin 20V/36V, and *trans*-vitisin B 30V.

### Statistical analysis

The data represent means  $\pm$  SD of three independent HSCCC experiments.

## RESULTS AND DISCUSSION

### HPLC analysis and selection of the two-phase solvent system

The most important step in the design of an HSCCC separation system is the selection of the solvent system.

Generally, two-phase solvent systems need to satisfy the following requirements: (1) the settling time of the solvent system should ideally be shorter than 30 s to ensure satisfactory retention of the stationary phase; (2) the partition coefficient ( $K$ ) of the target compounds should be  $0.5 \leq K \leq 2.0$  for efficient separation; (3) the separation factor between the two components ( $\alpha = K_2/K_1$ ,  $K_2 > K_1$ ) should be greater

than 1.5 (Lee *et al.*, 2011); and (4) the sample should not decompose or be denatured. In the current study, the  $K$  values of the four stilbenes were determined via HPLC, as described in the experimental section (Table 1). The solvent system was chosen based on the following observations: the *n*-hexane–ethyl acetate–methanol–water (1:2:1:2, v/v) solvent system provided suitable  $K$  values for the peaks,

TABLE 1

The partition coefficient ( $K$ ) of the target components in different solvent systems.

No.	Two-phase solvent system (v/v)	K value				Setting time (s)
		resveratrol	$\epsilon$ -viniferin	$\delta$ -viniferin	trans-vitisin B	
1	chloroform–methanol–water, 4:3:2	2.55	3.92	3.29	7.31	17
2	cyclohexane–ethyl acetate–methanol–water, 1:5:1:5	108.75	85	-	228.84	29
3	<i>n</i> -hexane–ethyl acetate–methanol–water, 2:5:2:5	9.76	26.4	22.92	59.34	33
4	<i>n</i> -hexane–ethyl acetate–methanol–water, 1:2:1:2	2.52	3.25	3.02	3.66	21
5	<i>n</i> -hexane–ethyl acetate–methanol–water, 2:5:1:2	6.79	10.5	11.71	17.39	29
6	ethyl acetate–ethanol–water, 25:1:25	-	92	-	-	20
7	ethyl acetate–ethanol–water, 5:1:5	-	102	-	-	19
8	chloroform-methanol- <i>n</i> -butyl alcohol- water 4:3:0.05:2	2.39	4.4	2.87	9.1	18

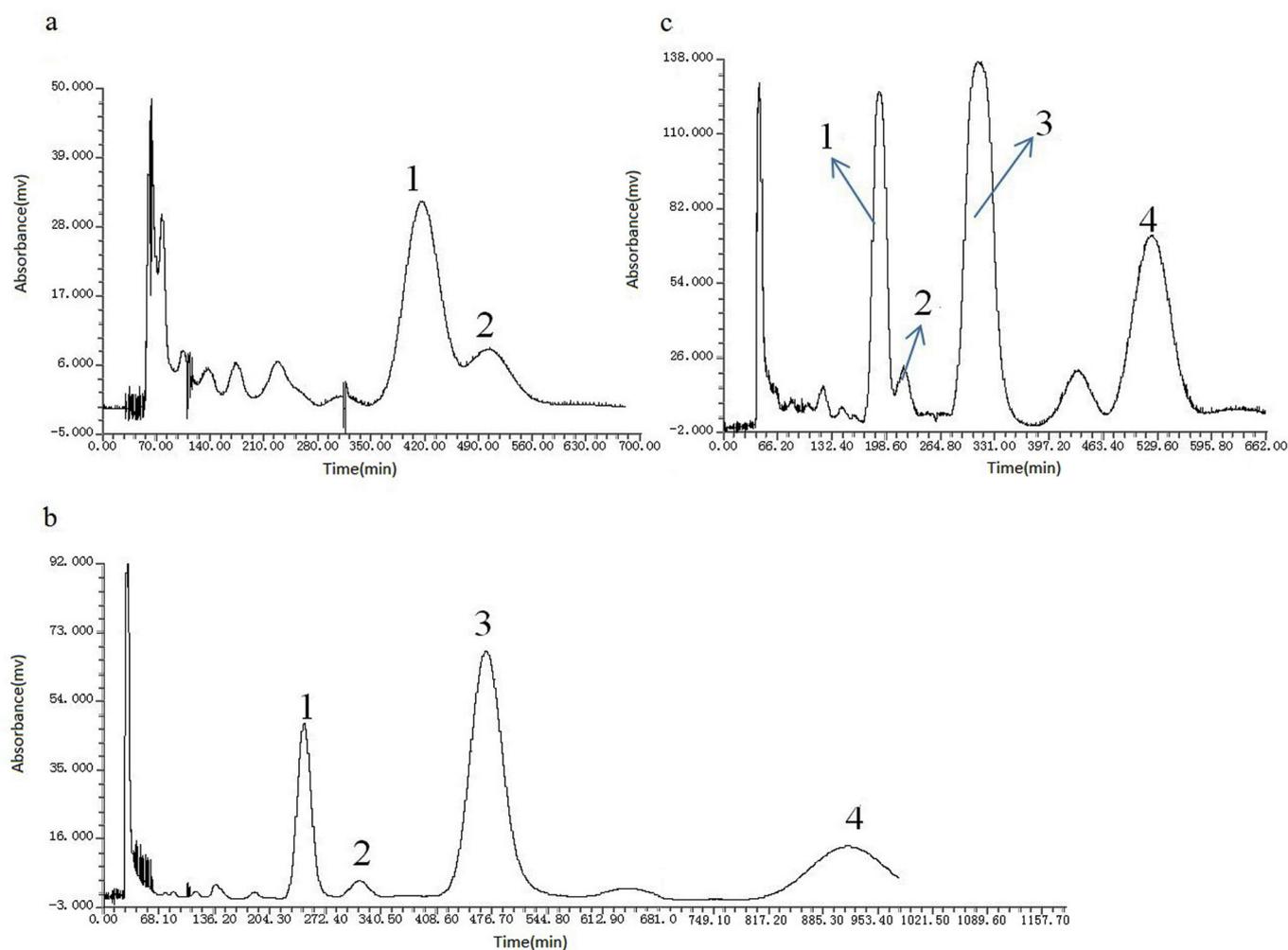


FIGURE 2

Preparative HSCCC separation of the crude sample from *Vitis vinifera* L. cv. Cabernet Sauvignon using (a) the two-phase solvent system composed of *n*-hexane–ethyl acetate–methanol–water (1:2:1:2, v/v); (b) the chloroform–methanol–water (4:3:2, v/v). The longest time recorded for the HSCCC procedure is 681 min, so after the completion of the first program, the second record immediately start without new sample; and (c) chloroform–methanol–*n*-butyl alcohol–water (4:3:0.05:2, v/v).

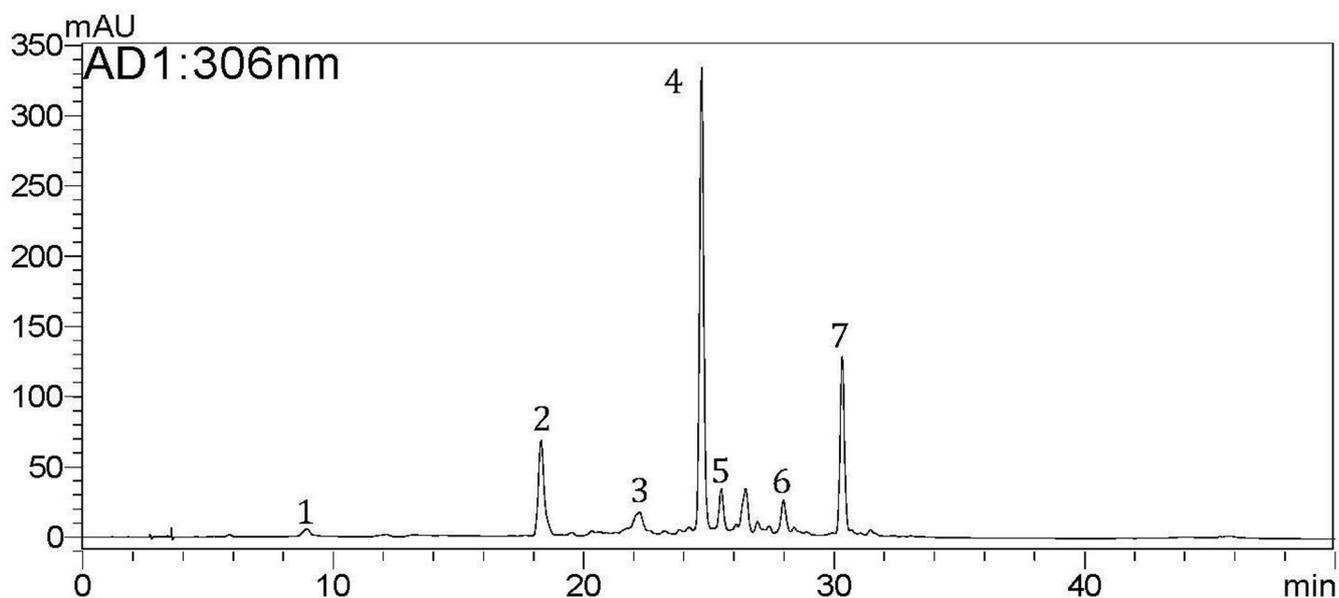


FIGURE 3

HPLC chromatograms of the crude extract from *Vitis vinifera* L. cv. Cabernet Sauvignon root. Peak 1 is piceid, 2 is *trans*-resveratrol, 3 is *cis*-resveratrol, 4 is compound 1, 5 compound 3, 6 is  $\delta$ -viniferin and 7 is compound 2.

which fulfilled requirement (2) mentioned above; however, this system provided a poor separation factor between these compounds. The chloroform–methanol–water (4:3:2, v/v) solvent system fulfilled requirement (3), but all the  $K$  values were higher than 2.0. Thus, the suggested *n*-hexane–ethyl acetate–methanol–water (1:2:1:2, v/v) (retention of the stationary was 61.43%) needed a shorter separation time, but it was unable to separate the stilbenes sufficiently (Fig. 2a). This system provided two peaks that contained two or more of the compounds. The chloroform–methanol–water system (4:3:2, v/v) (retention of the stationary phase was 60.71%) could separate the four stilbenes well, but it required separation times longer than 15 h (Fig. 2b). Thus, this system would be feasible if the separation time was shortened. The polar modifier *n*-butyl alcohol was added to the chloroform solvent system and formed a new solvent system consisting of chloroform–methanol–*n*-butyl alcohol–water (4:3:0.05:2, v/v). The new solvent system had a much shorter separation time and it sufficiently separated the four stilbenes, with good retention of the stationary phase (58.93%).

Figure 3 shows HPLC chromatograms of the crude extract from the roots. There are seven peaks. Peak 1 is piceid, 2 is *trans*-resveratrol, 3 *cis*-resveratrol, 4 compound 1; 5 compound 3, 6  $\delta$ -viniferin, and 7 is compound 2. Figure 2c shows the preparative HSCCC separation of 241 mg of crude extract using the solvent system composed of chloroform–methanol–*n*-butyl alcohol–water (4:3:0.05:2, v/v). Peak 1, from 172 min to 205 min, peak 2, from 211 min to 237 min, peak 3, from 280 min to 345 min, and peak 4, from 476 min to 569 min, were collected and concentrated. This separation yielded 7.1 mg  $\pm$  0.2 mg of peak 1, 1.1 mg  $\pm$  0.1 mg of peak 2, 18.7 mg  $\pm$  0.5 mg of peak 3, and 12.2 mg  $\pm$  0.2 mg of peak 4. Three independent experiments were carried out. Peaks 1, 2, 3 and 4 were identified as resveratrol,  $\delta$ -viniferin, compound 1 and compound 2, with purities of

97.89%, 90.61%, 94.37% and 78.38% respectively, based on the area percentage of the HPLC peak, as shown in Figs 4a to 4d. The lower purity of peak 4 was due to an unknown impurity. Peaks 3 and 4 from HSCCC separation were subjected to semi-preparative HPLC for further purification to obtain pure compounds. Two compounds were completely separated through the combined use of HSCCC and semi-preparative high-performance liquid chromatography, and the purities of these compounds were 97.39% and 95.35% respectively, as indicated by the HPLC analysis based on PDA detection (Fig. 4e). Piceid and compound 3 of the crude extract were not collected and detected through HSCCC.

#### Identification of compounds 1 and 2

The chemical structures of the peak fractions separated by HSCCC were identified according to their  $^1\text{H}$ -NMR and  $^{13}\text{C}$ -NMR data. Based on the comparison with the reference data, peak 3 (compound 1) and peak 4 (compound 2) were identified as  $\epsilon$ -viniferin and *trans*-vitisin B respectively. The results for each peak fraction were as follows.

Compound 1 (Peak 3 in Fig. 2c) was obtained as a pale yellow amorphous powder. ESI/MS/MS  $m/z$ : 455.1 [ $\text{M}+\text{H}$ ] $^+$  (calc. for  $\text{C}_{28}\text{H}_{23}\text{O}_6$ , 455.1), 453.1 [ $\text{M}+\text{H}$ ] $^-$  (calc. for  $\text{C}_{28}\text{H}_{21}\text{O}_6$ , 453.1), which was consistent with the molecular formula  $\text{C}_{28}\text{H}_{22}\text{O}_6$ , which corresponds to the resveratrol dimer. The  $^1\text{H}$ ,  $^{13}\text{C}$  NMR (600 MHz) data were as follows:  $^1\text{H}$  NMR  $\delta$ : 7.14 (2H, d,  $J = 6.8$  Hz), 7.04 (2H, d,  $J = 6.8$  Hz), 6.77 (2H, d,  $J = 6.8$  Hz), 6.64 (2H, d,  $J = 6.8$  Hz), 6.82 (1H, d,  $J = 16.3$  Hz), 6.57 (1H, d,  $J = 16.1$  Hz), 5.37 (1H, d,  $J = 4.9$  Hz), 4.35 (1H, d,  $J = 4.8$  Hz), 6.18 (3H, brs), 6.66 (1H, brs), 6.25 (1H, brs).  $^{13}\text{C}$  NMR  $\delta$ : 133.91, 127.9 (2C), 116.31 (2C), 158.54, 94.83, 58.31, 136.93, 107.49, 160.06, 102.22, 160.06, 107.49, 130.41, 128.78 (2C), 116.38 (2C), 158.39, 130.35, 123.73, 136.93, 120.06, 162.75, 96.86, 159.76, 104.37. The  $^{13}\text{C}$  NMR data corresponded well with  $\epsilon$ -viniferin, which has

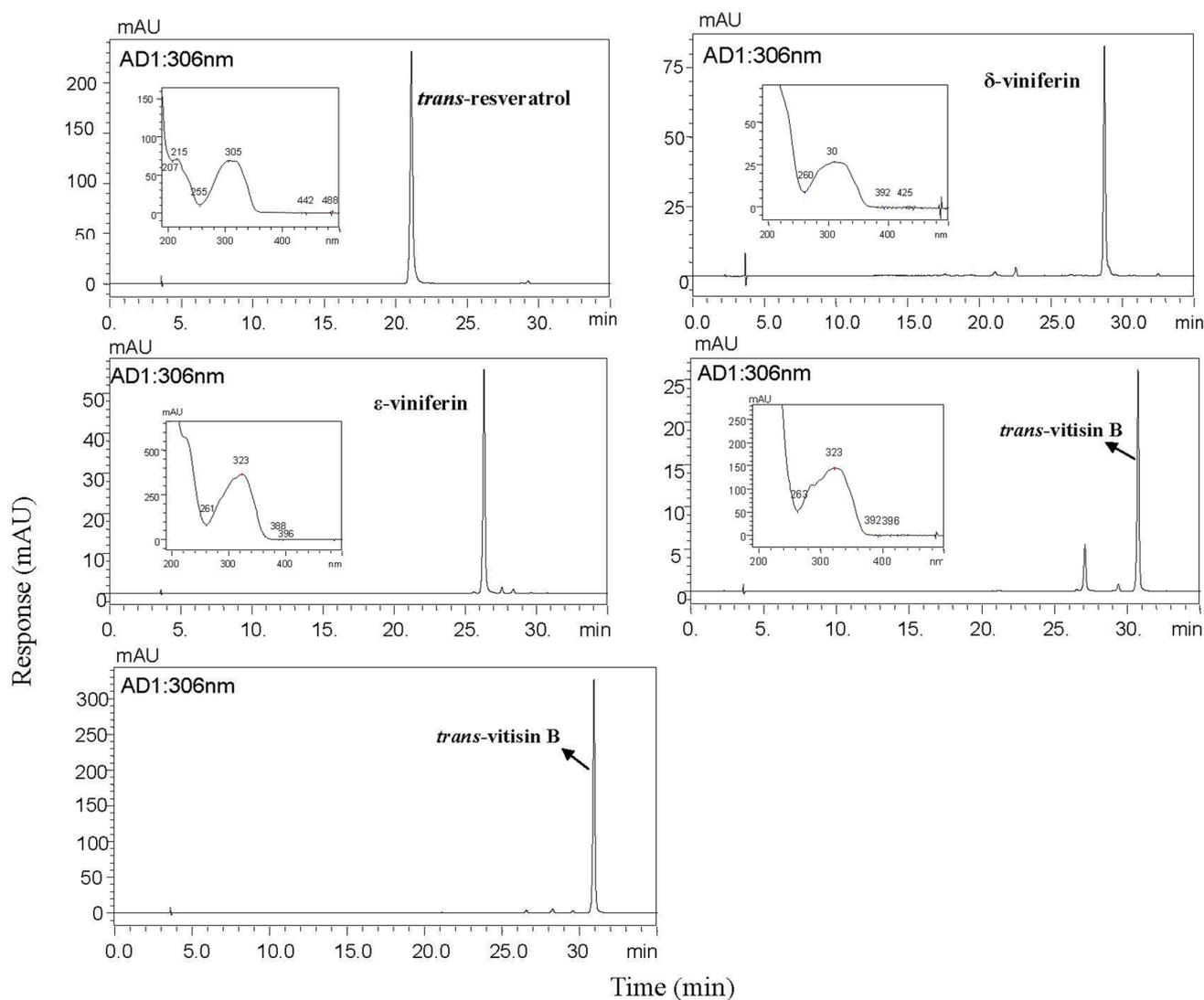


FIGURE 4

HPLC chromatograms and UV spectra of each separated peak. (a) HSCCC fraction of peak 1 in Fig. 2c, *trans*-resveratrol (97.89% purity); (b) HSCCC fraction of peak 2 in Fig. 2c,  $\delta$ -viniferin (90.61% purity); (c) HSCCC fraction of peak 3 in Fig. 2c,  $\epsilon$ -viniferin (94.37% purity); (d) HSCCC fraction of peak 4 in Fig. 2c, *trans*-vitisin B (78.38% purity). The collected fractions from HSCCC were analysed by HPLC using the following conditions: column, a PINNACLE II C18 column (250 mm  $\times$  4.6 mm, 5  $\mu$ m, Restek, USA); detection, 306 nm. (e) *trans*-Vitisin B was separated from fraction 4 via semipreparative HPLC (95.35% purity).

been reported in the literature (Li *et al.*, 1996). However, the  $^1\text{H}$  NMR chemical shifts were not entirely consistent with  $\epsilon$ -viniferin, as a 0.06 ppm smaller downfield shift was observed for almost all of the data. In addition, the coupling constants relevant to the molecular structure were consistent with  $\epsilon$ -viniferin. The spectrum exhibited 15 proton signals in the olefinic proton region ( $\delta = 4.5$  to 8.5), among which a pair of doublet signals at  $\delta$  6.82 and  $\delta$  6.57 were assigned to a pair of *trans*-olefinic protons, as shown by the coupling constants ( $J = 16.3$  Hz, 16.1 Hz). Hydrogen chemical shift deviations may result from the different solvents.  $\text{CD}_3\text{OCD}_3$  was used in the literature, whereas  $\text{CD}_3\text{OD}$  was used in this paper. Active hydrogen is susceptible to the effects of  $\text{CD}_3\text{OD}$ , hence the changes in the chemical shifts. In addition, the UV spectra of compound 1 corresponded to  $\epsilon$ -viniferin (Amira-Guebailia *et al.*, 2009; Kong *et al.*, 2011)

Compound 2 (Peak 4 in Fig. 2c) was obtained as pale amorphous powder. ESI/MS/MS  $m/z$ : 907.2  $[\text{M}+\text{H}]^+$  (calc. for  $\text{C}_{56}\text{H}_{43}\text{O}_{12}$ , 907.2), 905.2  $[\text{M}+\text{H}]^-$  (calc. for  $\text{C}_{56}\text{H}_{42}\text{O}_{12}$ , 905.2), which is consistent with the molecular formula  $\text{C}_{56}\text{H}_{43}\text{O}_{12}$ , which corresponds to the resveratrol tetramer. The  $^1\text{H}$ ,  $^{13}\text{C}$  NMR (600 MHz) data were as follows:  $^1\text{H}$  NMR  $\delta$ : 7.18 (2H, d,  $J = 8.4$  Hz), 7.14 (2H, d,  $J = 8.4$  Hz), 6.99 (1H, dd,  $J = 8.2, 1.8$  Hz), 6.83 (2H, d,  $J = 8.4$  Hz), 6.77 (2H, d,  $J = 8.2$  Hz), 6.66 (1H, brs), 6.69 (1H, brs), 6.68 (1H, d,  $J = 15.12$  Hz), 6.58 (1H, brs), 6.51 (1H, d,  $J = 14.7$  Hz), 6.53 (2H, d,  $J = 6.6$  Hz), 6.59 (2H, d,  $J = 8.8$  Hz), 6.28 (1H, d,  $J = 2.2$  Hz), 6.25 (1H, d,  $J = 2.1$  Hz), 6.10 (1H, d,  $J = 2.2$  Hz), 6.14 (3H, brs), 6.07 (1H, t), 5.98 (2H, d,  $J = 2.2$  Hz), 5.43 (1H, d,  $J = 5.1$  Hz), 5.37 (1H, d,  $J = 6.6$  Hz), 5.34 (1H, d,  $J = 4.6$  Hz), 4.36 (1H, d,  $J = 4.5$  Hz), 4.34 (1H, d,  $J = 6.5$  Hz), 4.25 (1H, d,  $J = 5.0$  Hz).  $^{13}\text{C}$  NMR  $\delta$ : 127.78 (2C),

128.2 (2C), 126.4, 116.31 (2C), 116.49 (2C), 125.54, 110.69, 130.49, 104.62, 124.2, 116.03 (2C), 127.78 (2C), 96.5, 96.91, 107.49, 102.28, 102.51, 107.49, 107.49, 107 (2C), 92.25, 94.78, 94.72, 58.24, 57.93, 52.99. Based on these data, compound 2 was confirmed as *trans*-vitisin B, which has been reported in the literature (Oshima *et al.*, 1995).

## CONCLUSIONS

In the present study, four similar stilbenes were successfully isolated for the first time from the roots of *Vitis vinifera* L. Cabernet Sauvignon, which has a higher stilbene level than wine of high purity, using conventional HSCCC. The crude extract of root was separated and purified directly by HSCCC using chloroform–methanol–*n*-butyl alcohol–water (4:3:0.05:2, v/v) solvent system. This study was the first to separate  $\delta$ -viniiferin,  $\epsilon$ -viniiferin and *trans*-vitisin B by using a single HSCCC method. The results of our study demonstrate that the method is a feasible, economical and efficient technique for the rapid preparative isolation of bioactive components from plant materials.

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